

# Database



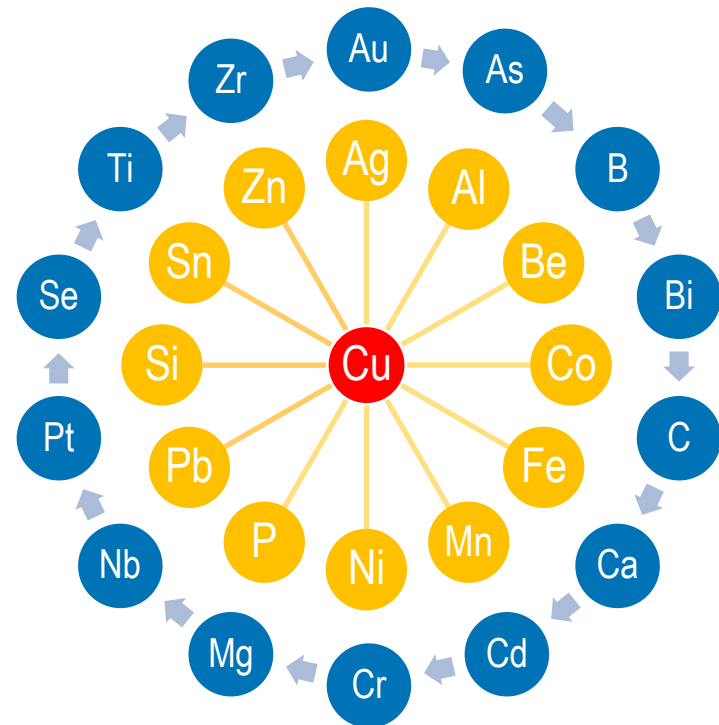
Qing Chen

Computational Thermodynamics and Kinetics Seminar  
June 16, 2016, Stockholm, Sweden

- **New and updated databases**
  - TCCU1
  - MOBCU1
  - SLAG4
  - SSOL6
  - TCNI6, 7, 8
  - Demo-databases
- **On-going database development**
  - TCFE9
  - TCNI9
  - TCOX7
  - TCTI1

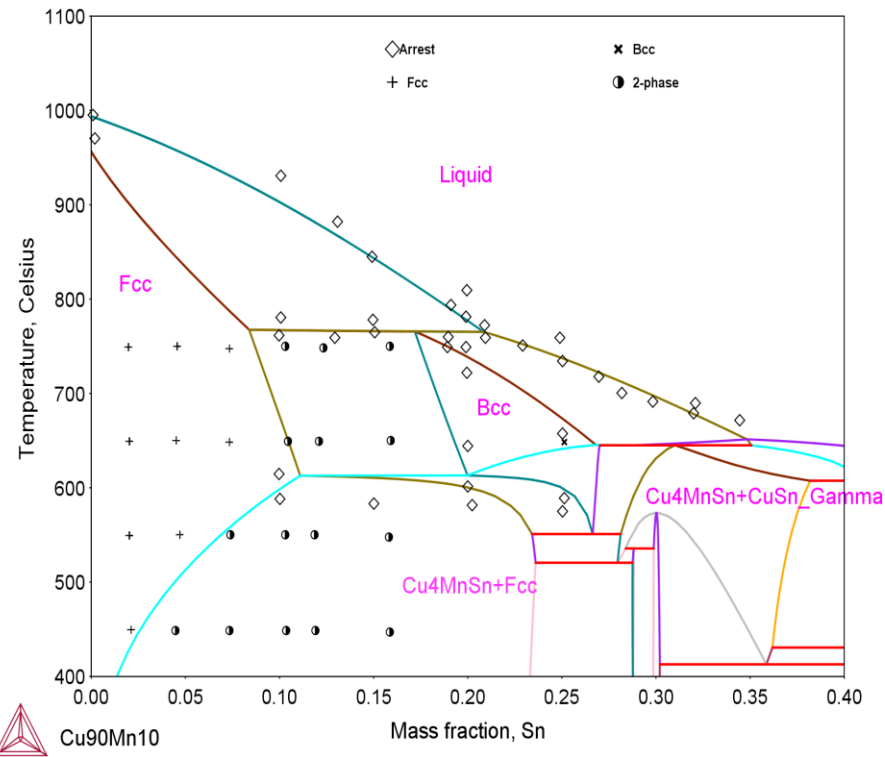
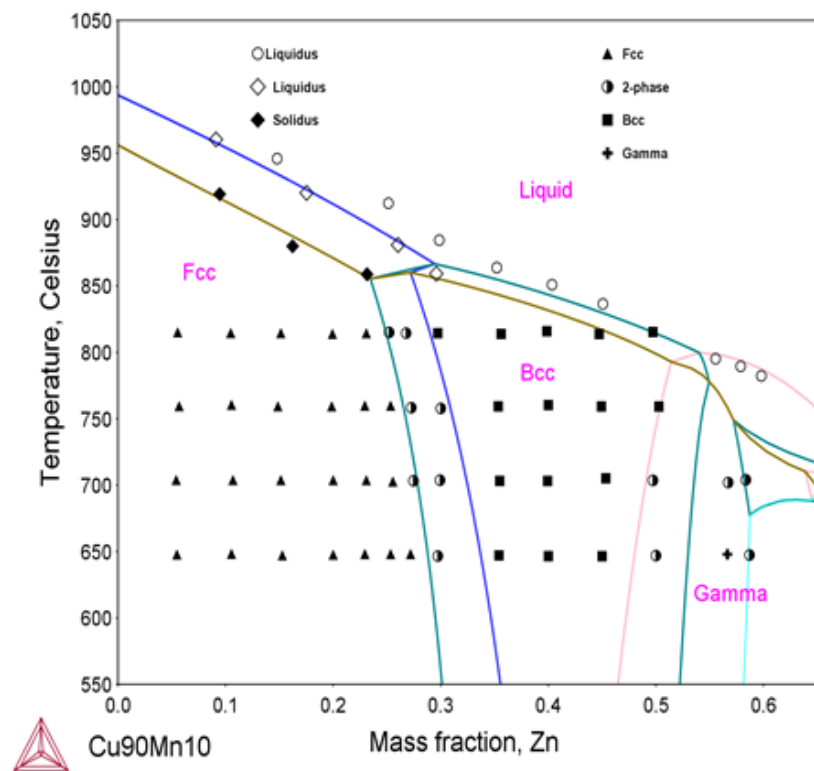
## Thermodynamic database for Cu-based alloys

- 27 elements
- 113 assessed binary systems
- 40 assessed ternary systems
- 219 phases

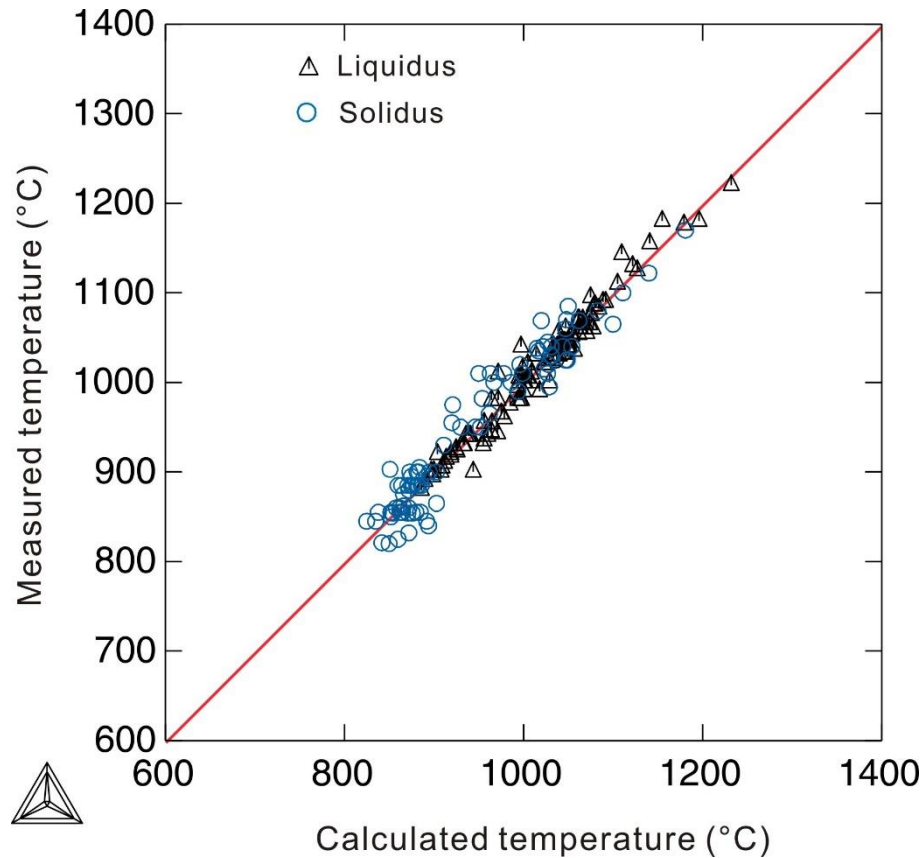


## Cu-Mn-Zn system

## Cu-Mn-Sn system



Calculated vertical sections of 10wt.% Mn in (a) Cu-Mn-Zn and (b) Cu-Mn-Sn systems along with experimental data.

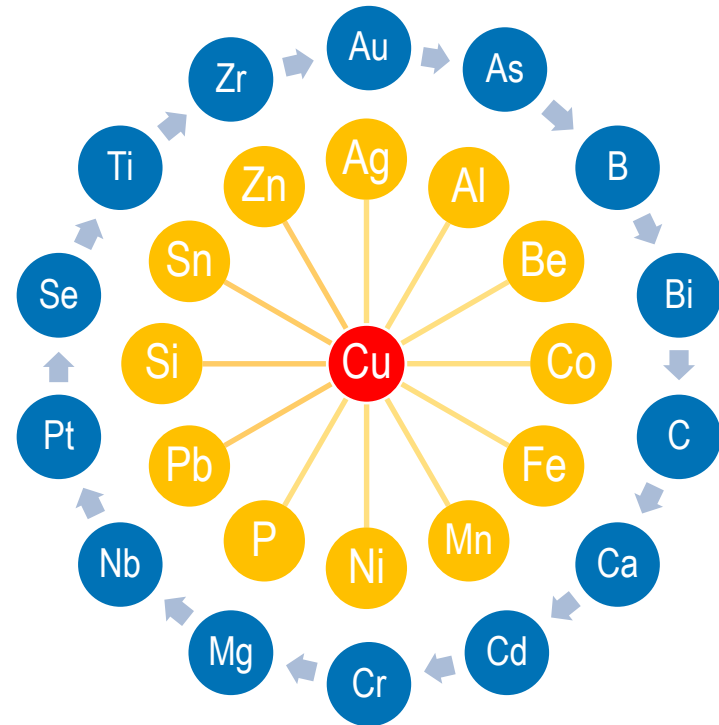


105 multi-component industry alloys

- Cu-Al-Fe-Ni based alloys
- Cu-Fe-Ni-Mn based alloys
- Cu-Pb-Sn-Zn based alloys
- Cu-Al-Si based alloys
- Cu-Fe-P based alloys
- Cu-Be-Co based alloys
- .....

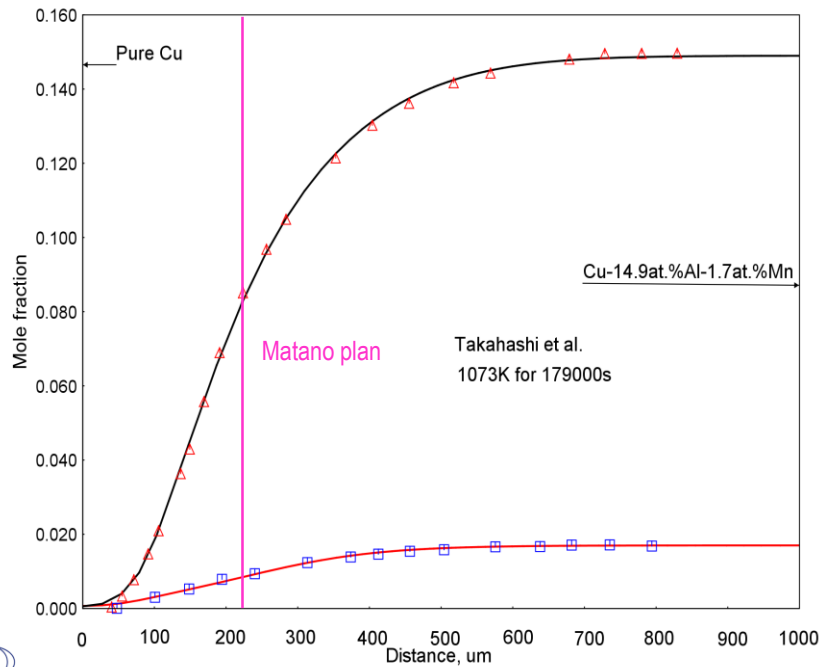
## Atomic mobility database for Cu-based alloys

- 27 elements
- Liquid and fcc phases
- 37 binary systems
- 26 ternary systems
- 1 quaternary systems

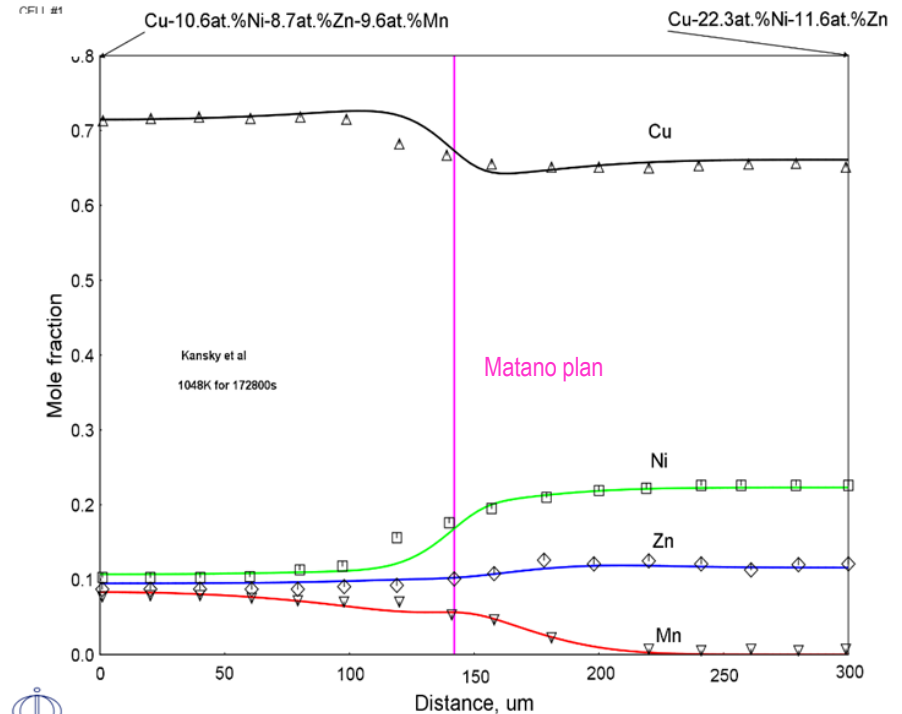


## □ Diffusivity in solid (fcc) phase

### Cu-Al-Mn system



### Cu-Ni-Mn-Zn system



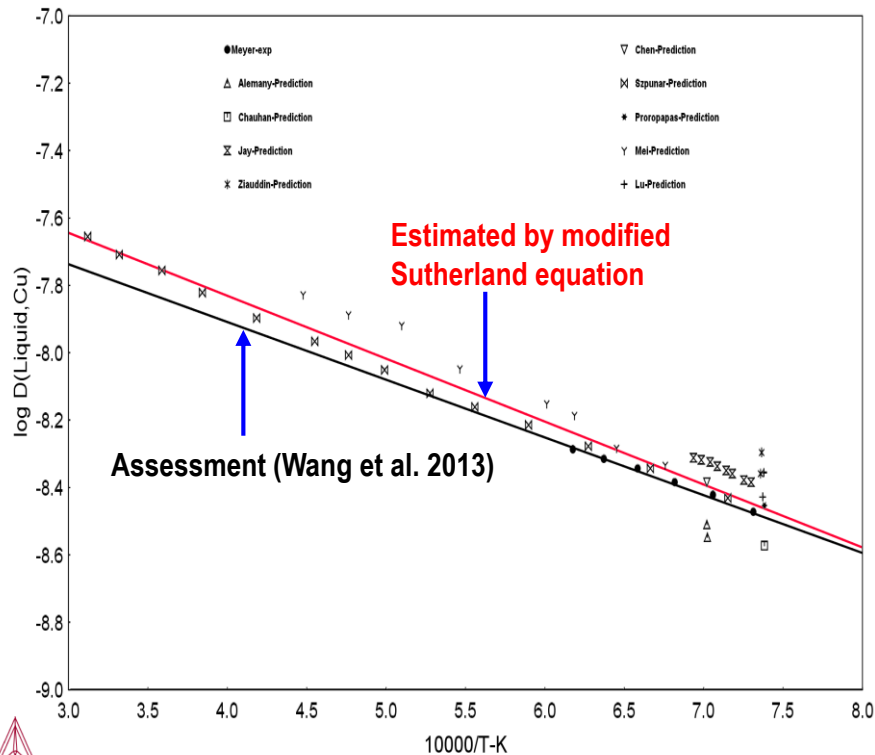
Calculated concentration profile compared with experimental data (a) Cu//Cu-14.9Al-1.7Mn diffusion couple, (b) Cu-10.6Ni-8.7Zn-9.6Mn//Cu-22.3Ni-11.6Zn diffusion couple

Experiments from Takahashi et al. , Defect. Diffus. Forum. 95 (1993) 641

Kansky et al. Metall. Mater. Trans. A 16 (1985) 1123

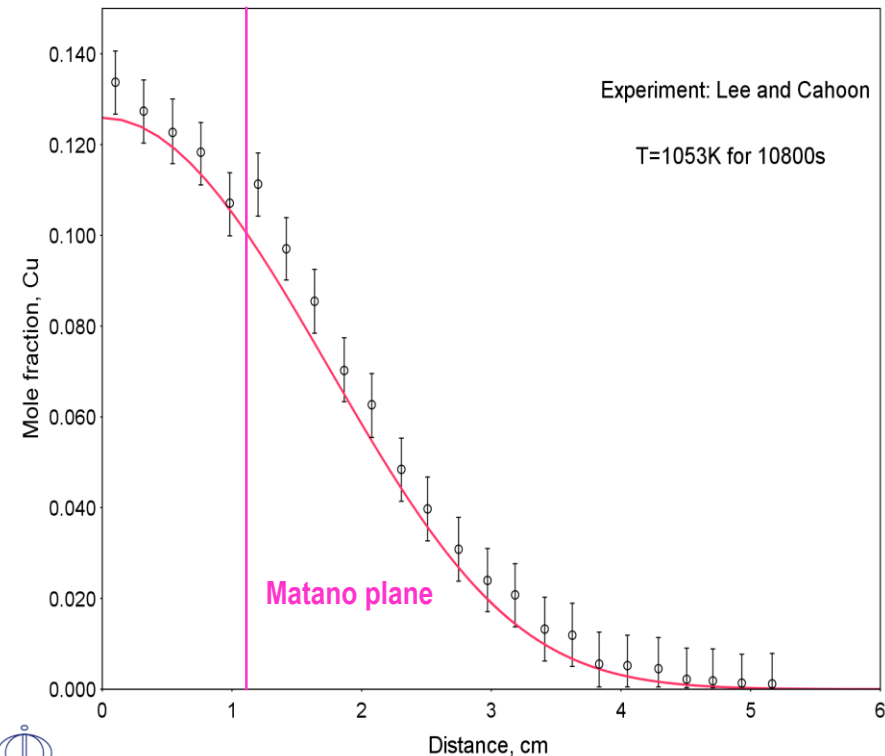
## Diffusivity in liquid phase

### Self-diffusivity of liquid Cu



Calculated self-diffusivity of Cu in liquid phase compared with experimental data.

### Cu-Al system

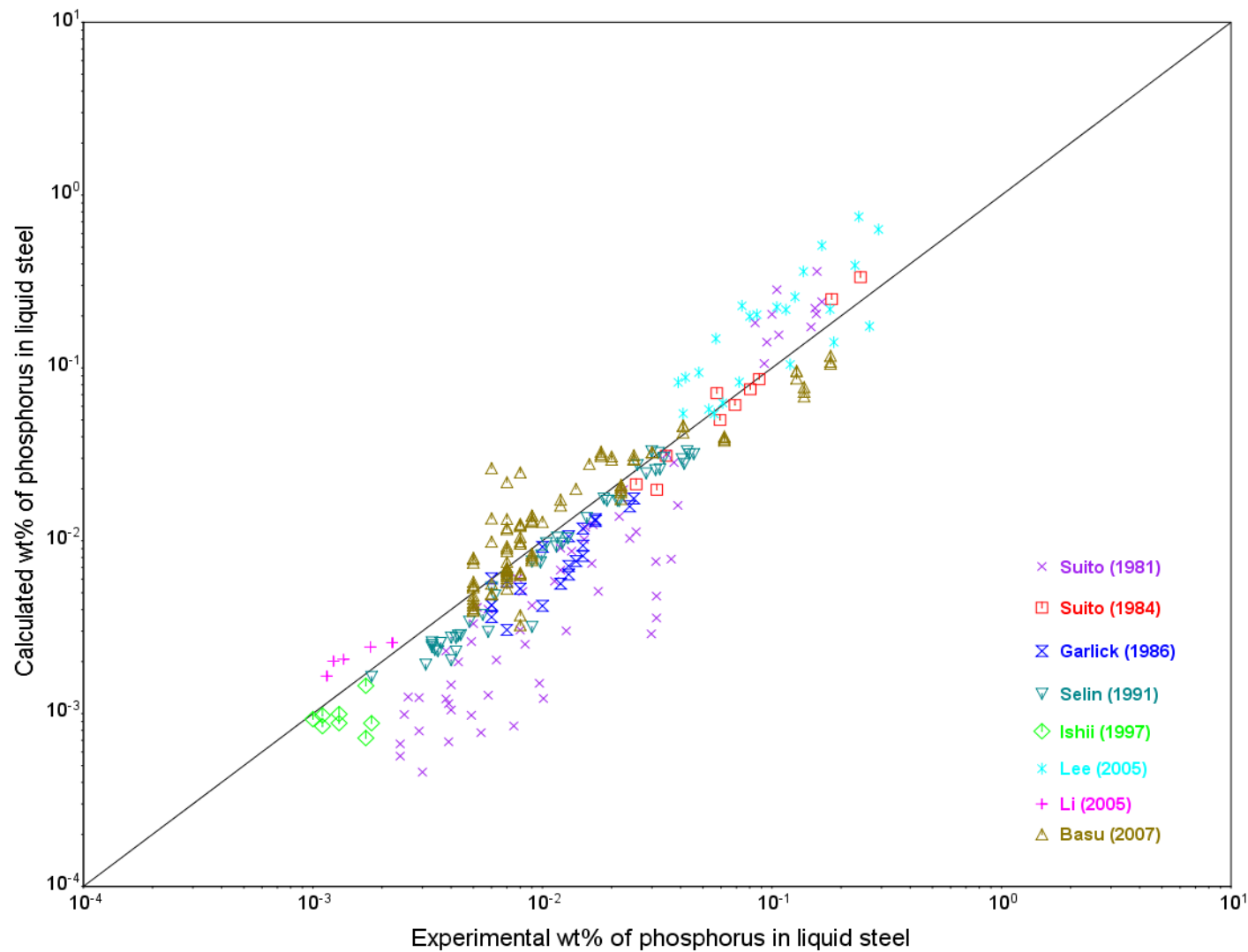


Calculated concentration profile compared with experimental data Al-0.231Cu // Al diffusion couple annealed at 1053 K for 10800 s.



- **P**hosphorus was added.
- The gas phase was simplified to accelerate the equilibrium calculation.
- Better description of some solid fluorides such as  $\text{AlF}_3$ ,  $\text{FeF}_2$ ,  $\text{FeF}_3$ ,  $\text{MgF}_2$ , and  $\text{MnF}_2$ .
- Oxygen solubility in liquid iron and  $\text{FeO}$  stability corrected.

## Phosphorus solubility in liquid iron in equilibrium with steelmaking slag



SSOL6 contains assessed thermodynamic data from the literature and provided by SGTE members. SSOL6 includes 729 alloy systems consisting of:

- 575 binary systems
- 133 ternary systems
- 20 higher-order systems
- 1331 phases.

The key changes from SSOL5 to SSOL6 are listed below:

- Hydrogen (H) is added to the database – now 79 elements.
- Many binary and ternary systems are added or updated in SSOL6.

- The TCNI6, TCNI7, TCNI8 databases have been slenderized.
  - Loading the databases is now 2 times faster.
  - Saving calculation time by as high as 50%.

## New DEMO-databases

- NIDEMO: Includes Ni, Al and Cr (full ternary description)
- MNIDEMO: Includes mobility data for Ni, Al and Cr in BCC\_B2 and FCC\_L12
- MALDEMO: Includes mobility data for Al, Cu, Sc and Si in FCC\_A1.

## Updated DEMO-databases

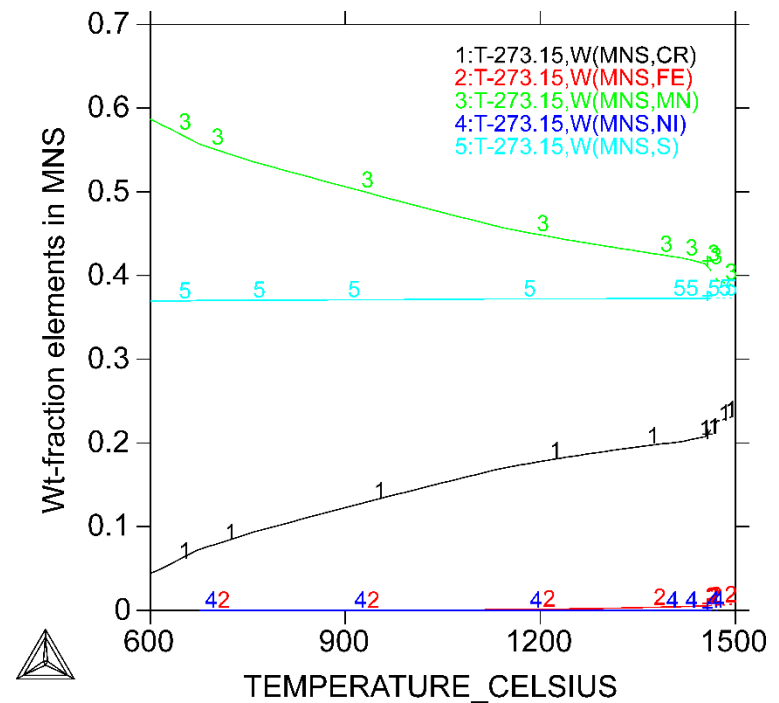
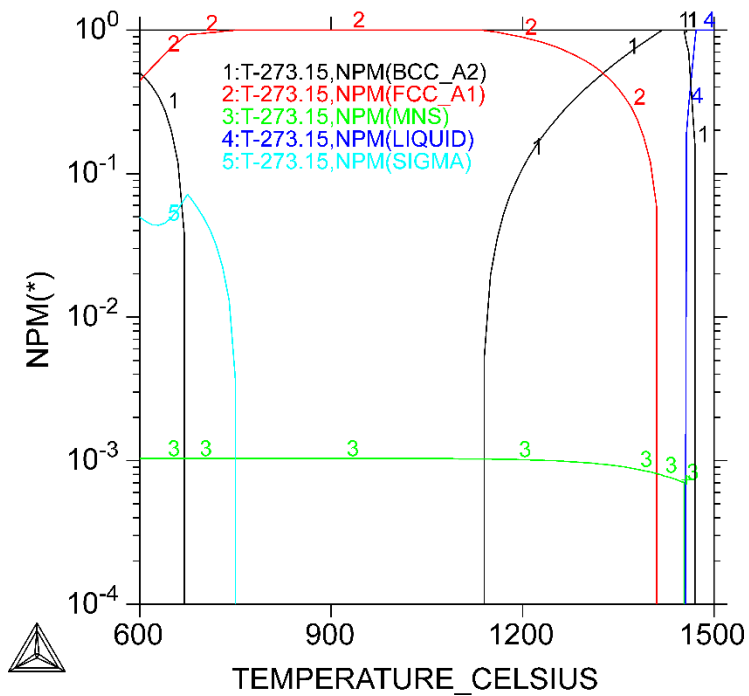
- ALDEMO: Includes Al, Cu, Sc and Si. **Sc has been added to the database.**

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  - SSOL6
  - TCNI6, 7, 8
  - Demo-databases
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  - TCTI1

- Ongoing work:
  - Improved sulfides description.
  - Improved Laves phase prediction.
  - Improved Sigma phase prediction.
  - Improved Fe-Cu-X systems.
  - Improved high Mn-high Al systems.
  - Extended Kappa phase description.
  - Improved ordered phases BCC\_B2 and FCC\_L12 descriptions.
  - Improved Borides description.
  - To include Ce and Gd

Improved sulfides description.

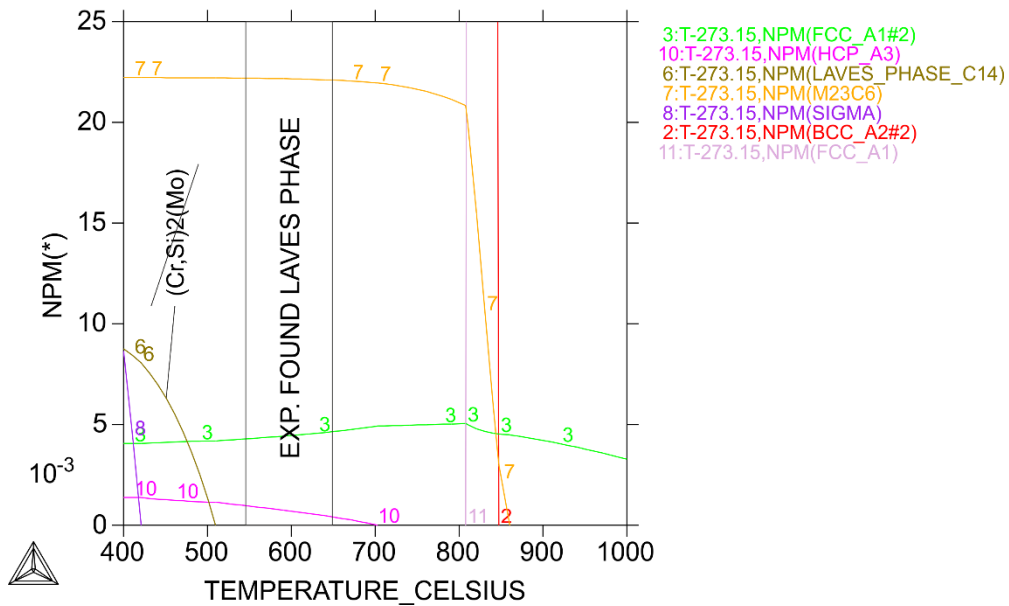
DATABASE:TCFE9  
Fe-18Cr-8Ni-1Mn-0.03S



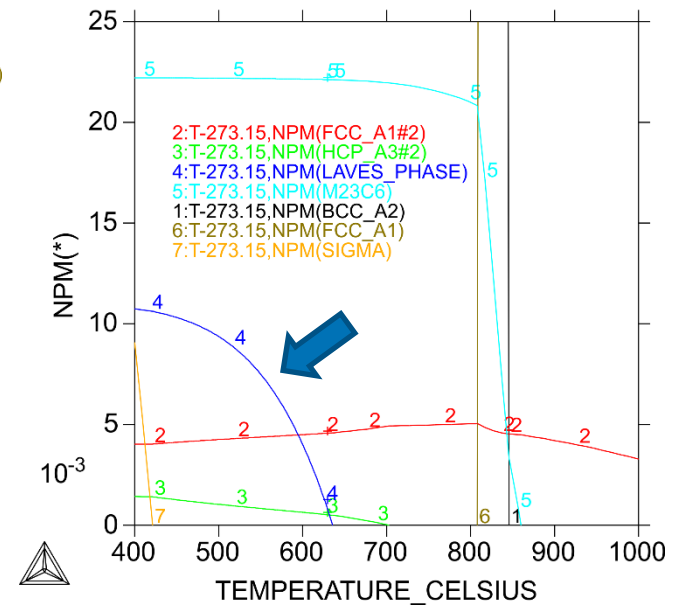


## Improved Laves phase prediction.

DATABASE:TCFE7



DATABASE:TCFE9



	C	Cr	Mn	Mo	N	Nb	Ni	V	Si
ASME91	0.1	8	0.5	1	0.06	0.05	0.1	0.2	0.4

- *Equilibria at liquidus - solidus*
- *Equilibria of matrix phases at high temperature*
- *Equilibria of secondary phases at intermediate temperature*

## Database modifications

### Binary

Cu-N Fcc	Ni-N Bcc	Si-N Fcc, Bcc	Mn-Ni Fcc, bcc, liq
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### Ternary

Cr-Fe-Mn Fcc	Cr-Fe-Mo Sigma, Laves	Cr-Fe-N Bcc, Fcc	Cr-Fe-Si Sigma	Cr-Mn-N Bcc	Cr-Ni-N Bcc, Fcc
Fe-Mn-Mo Bcc, Fcc, Laves	Fe-Mn-N Bcc	Fe-Mo-Si Sigma, Laves	Fe-Ni-N Bcc	Fe-W-N Bcc	Fe-Mo-V Laves
Fe-Mo-Nb Laves	Fe-Ni-Mo Laves	Cu-Fe-N	Fe-Si-N	Fe-Mn-Ni Fcc, bcc, liq	

### Quaternary

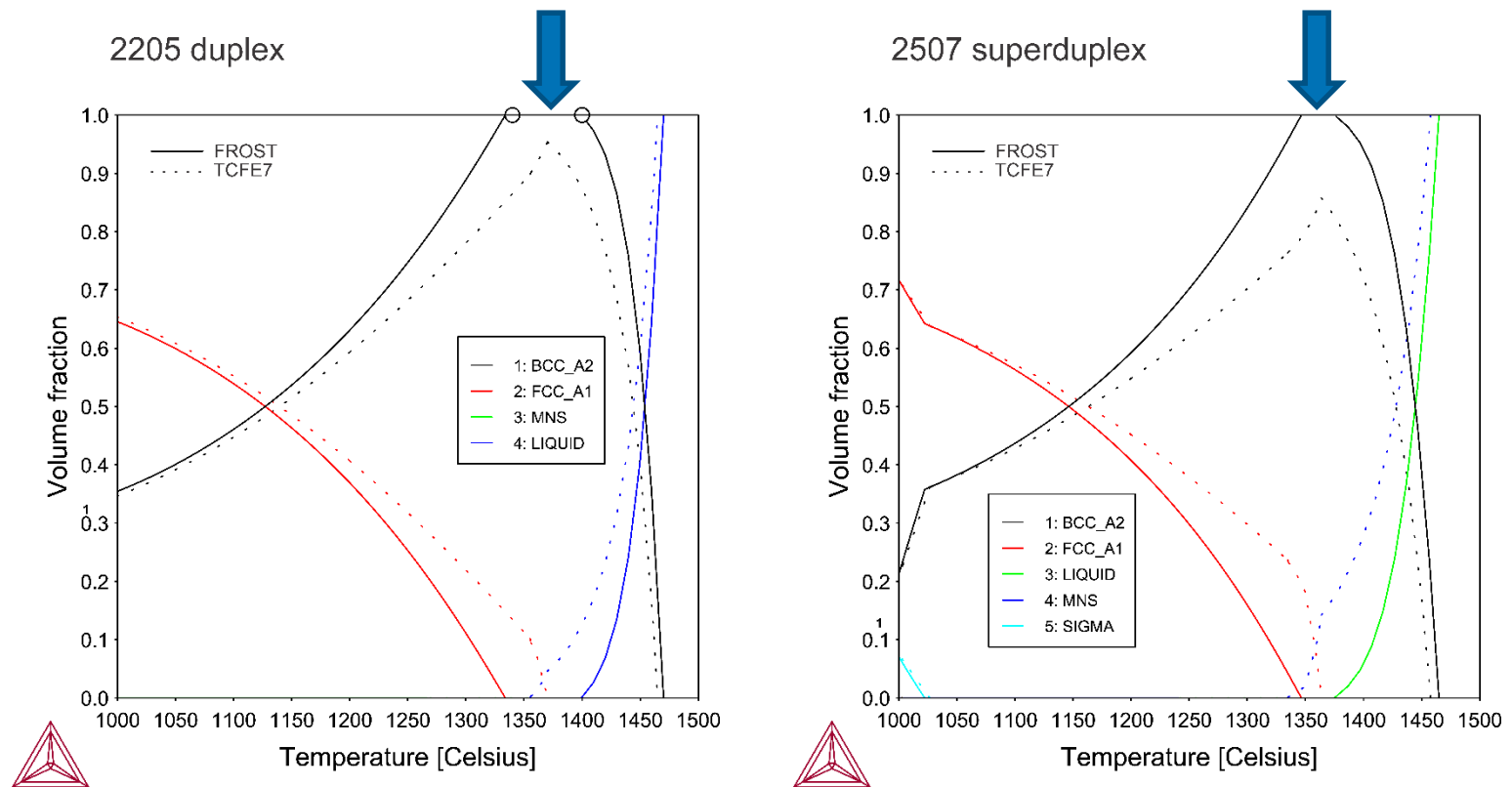
Cr-Fe-Mn-N Fcc, Hcp	Cr-Fe-Ni-N
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### Quinary

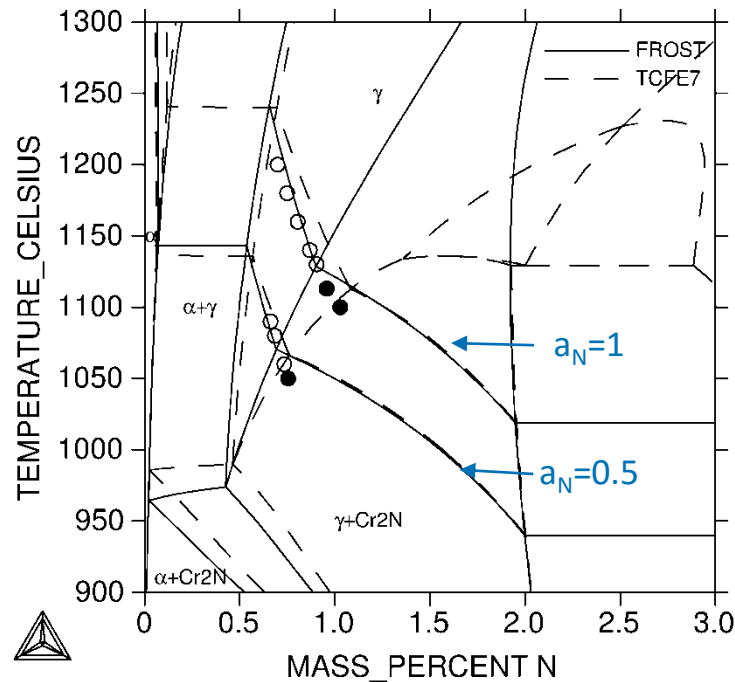
Cr-Fe-Mo-Ni-N

*A: Equilibria at liquidus - solidus*

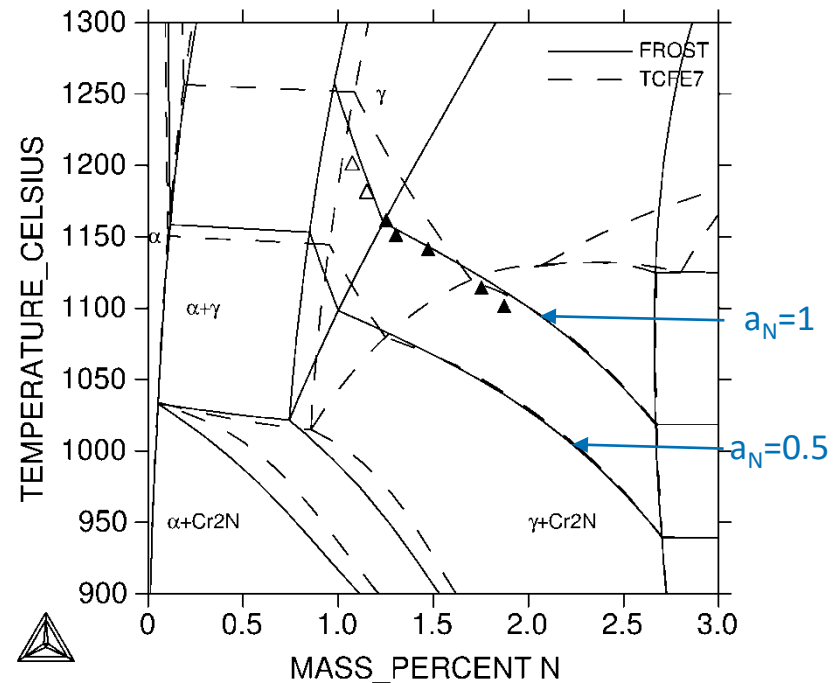
Phase fractions vs. temperature: 2205 duplex & 2507 superduplex



*C: Equilibria of secondary phases at intermediate temperature*



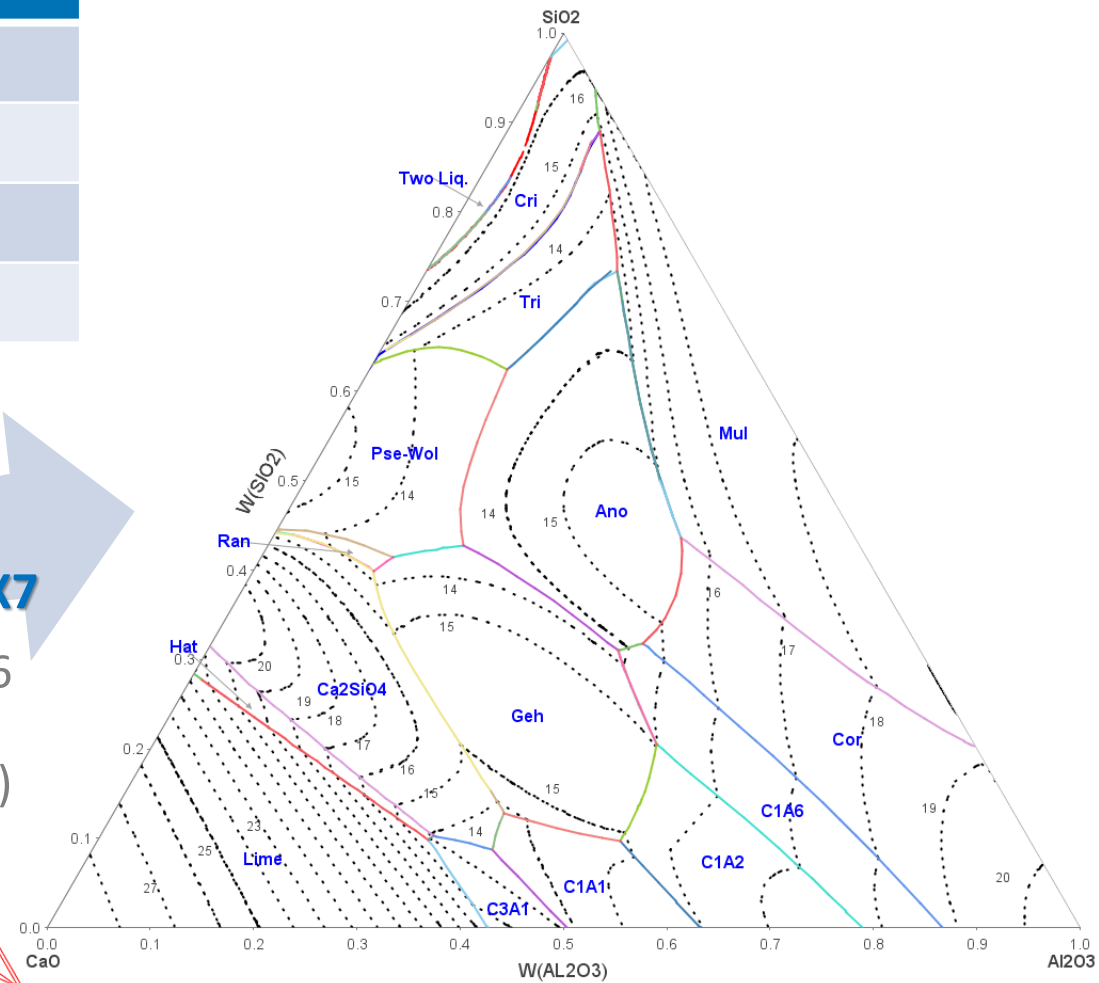
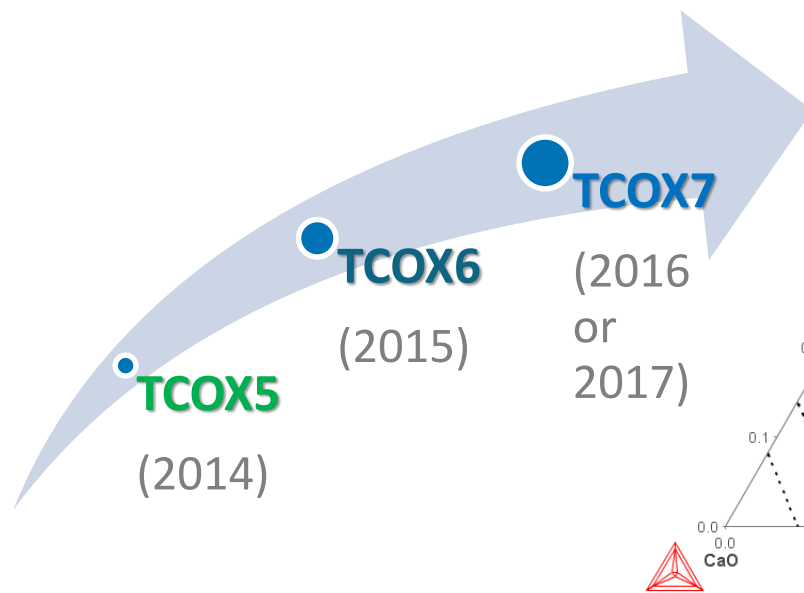
Fe-19.8Cr-0.11Mn-N

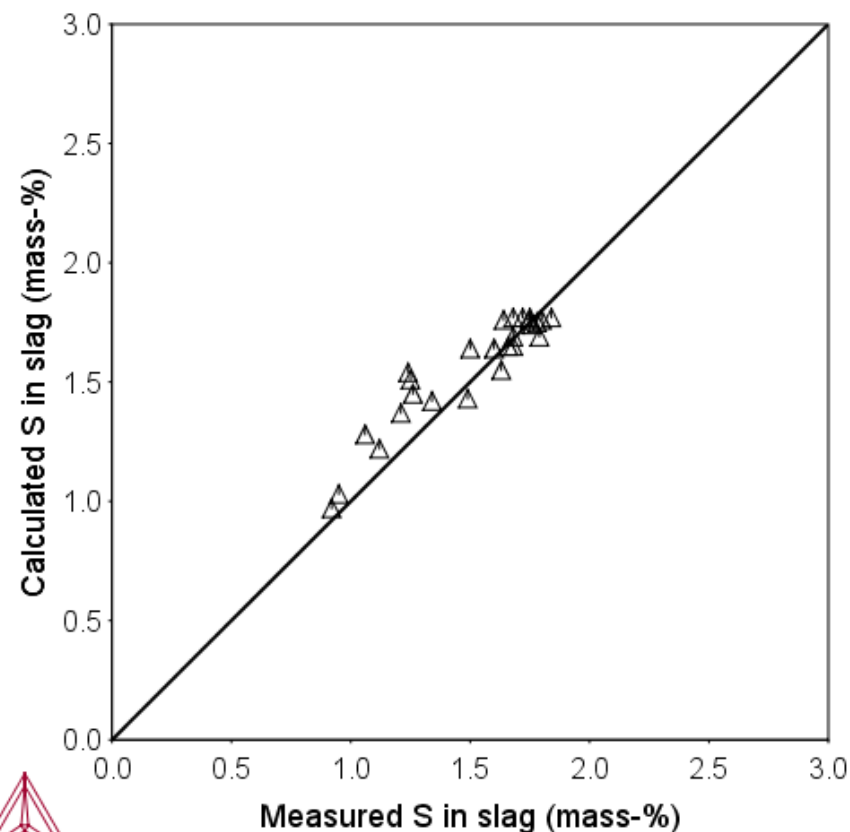
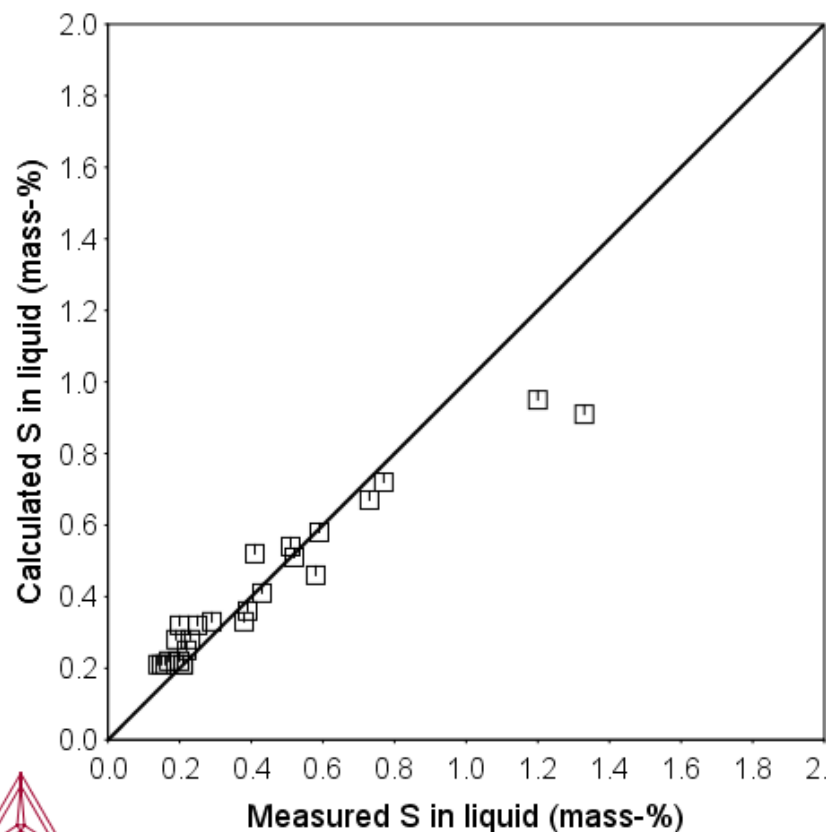


Fe-24.3Cr-0.44Mn-N

Temperature [°C] vs. nitrogen content [wt-%] with isoactivity lines  $a_N=0.5$  and 1 and experimental results.

Al	C	Ca	Cr
Fe	Mg	Mn	Ni
O	Si	Y	Zr
F	Nb	S	Ti
Cu	La	Gd	

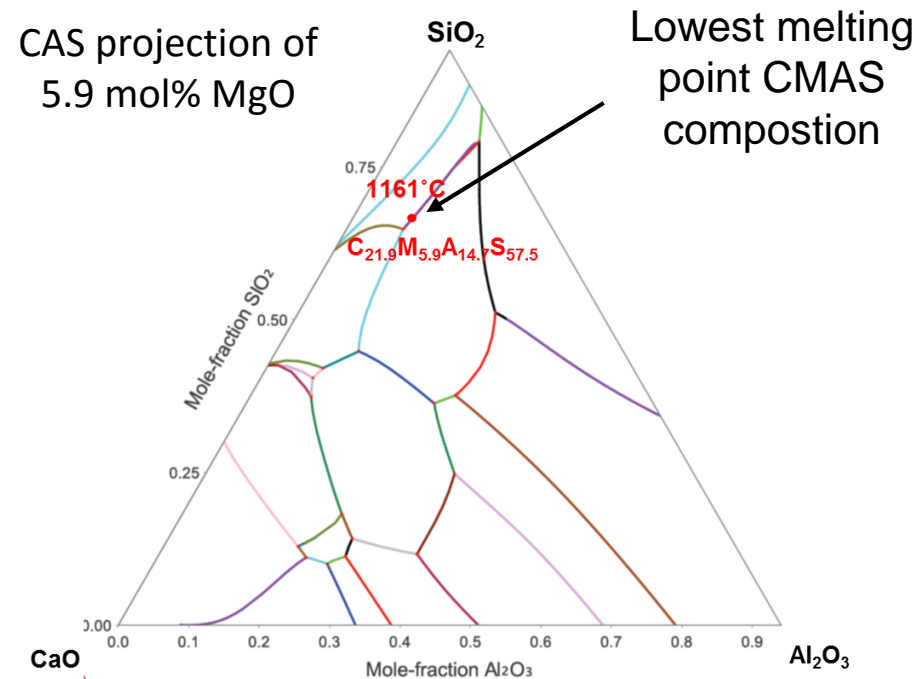
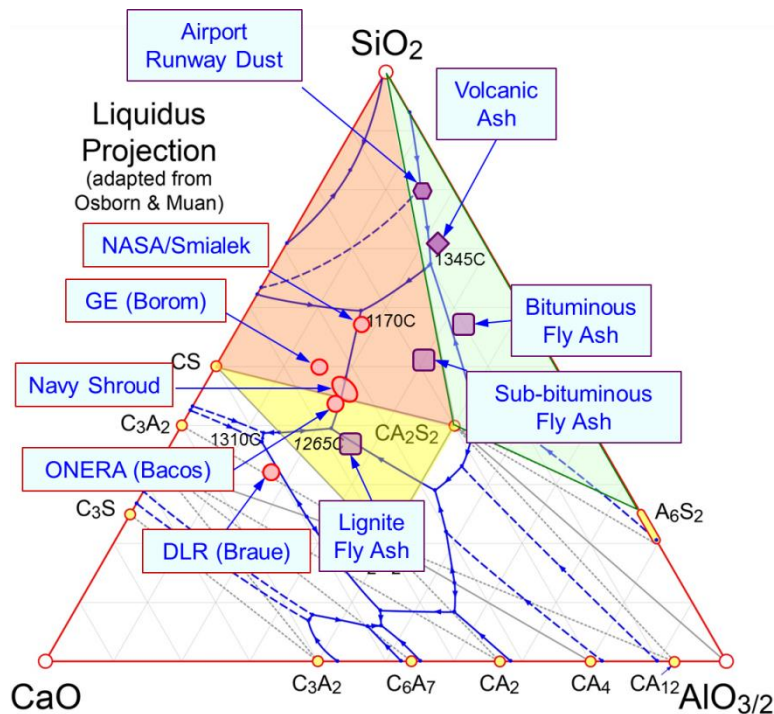




Computed Sulfur content in Cu liquid and slags in Cu-Al-Ca-Mg-Si-O-S system by mixing (liquid) Cu,  $\text{Cu}_2\text{S}$  and a CMAS slag.  
Experimental data are from C. Allertz's thesis (2016).

## Lowest melting point CMAS

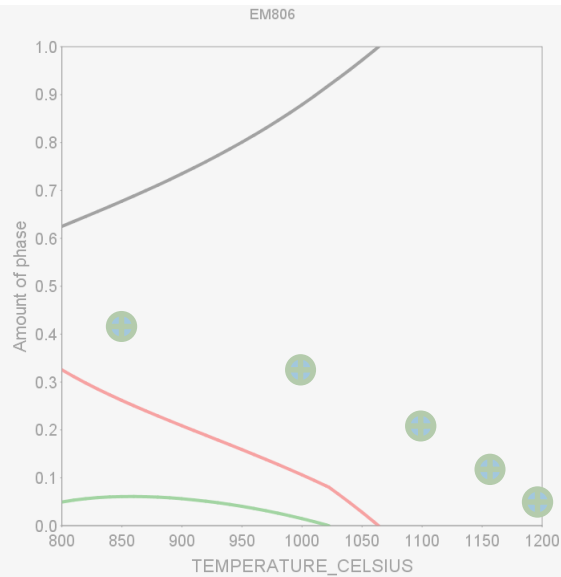
- NASA/Smialek composition ( $C_{23}M_9A_{13}S_{55}$ ) melts at  $1170^\circ\text{C}$
- Database predicted lowest melting point at  $1161^\circ\text{C}$  CMAS composition:  
 $C_{21.9}M_{5.9}A_{14.7}S_{57.5}$



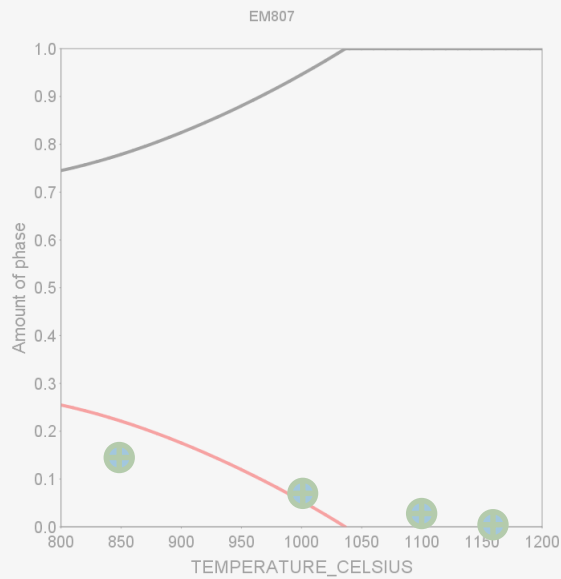
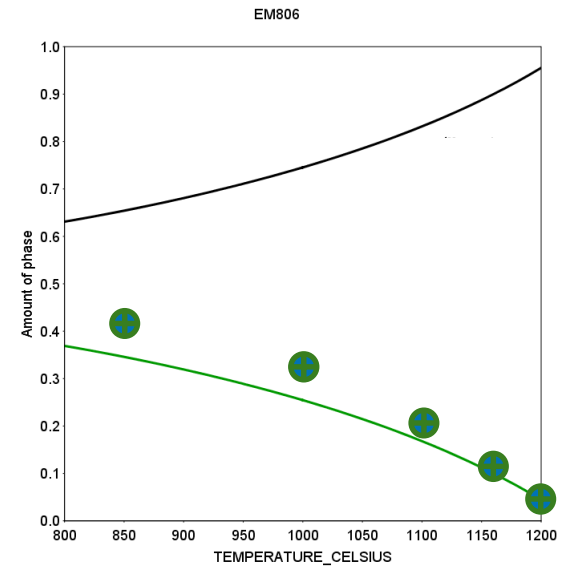


Experimental investigation in the Ni-Al-Co-Fe-Nb-Ti system was performed.

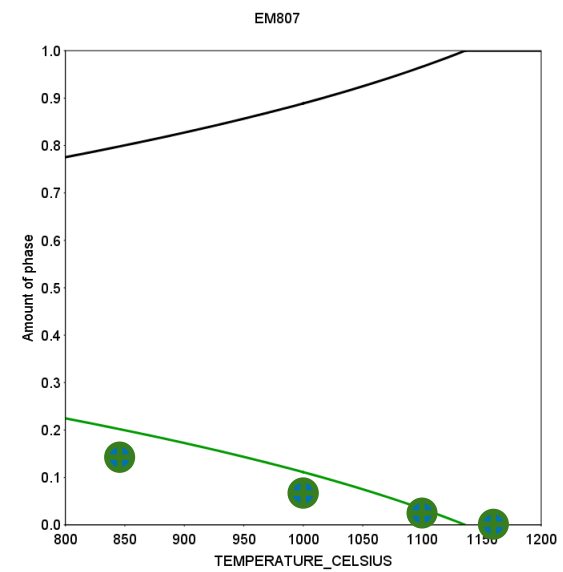
- All alloys showed high amount of eta-phase while TCNI8 predicts gamma-prime.
- Additional ternary and quaternary alloys were produced, heat-treated and analyzed for the TCNI9 development.
- After adjusting TCNI9 to this new experimental info, eta phase is predicted in the Ni-Al-Co-Fe-Nb-Ti alloys.



γ  
γ'  
η

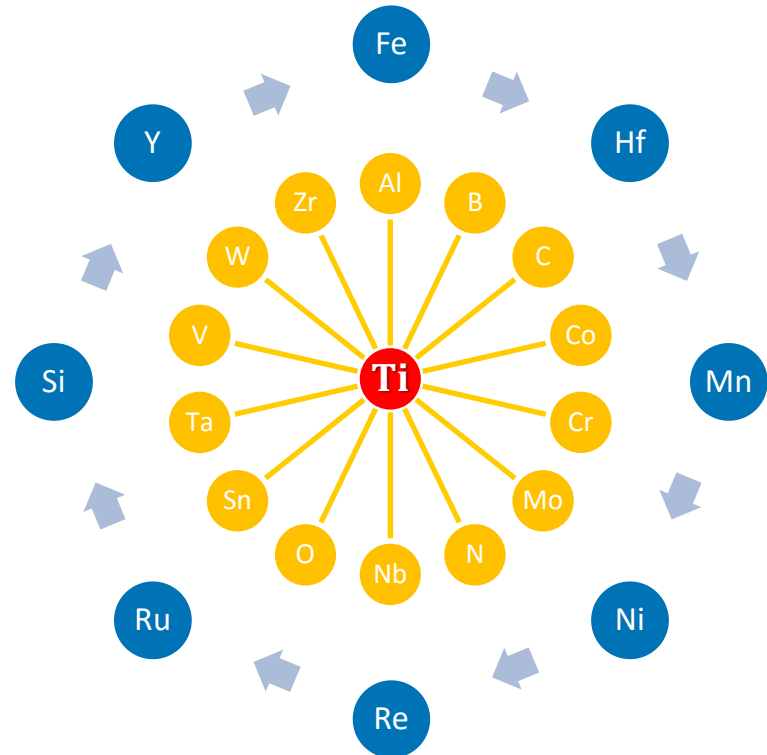


γ  
γ'  
η



- ❑ 23 elements
- ❑ 237 binary systems
- ❑ 65 ternary systems
  - 1 directly added
  - 29 to be reassessed
  - 35 to be assessed
  - 3 to be extrapolated

All important phases  
including  $\omega$  phase and  
its variants



## DFT calculations

### $\omega$ , pure elements

- Al, B, C, Co, Cr, Cu, Fe, Hf, Mn, Mo, Nb, Ni, Pd, Pt, Re, Ru, Si, Ta, Ti, V, W, Y, Zr

### $\omega$ , ordered structures, C32 and B8<sub>2</sub>, Ti<sub>2</sub>X + TiX<sub>2</sub>

- X = Al, Co, Cr, Cu, Fe, Hf, Mn, Mo, Nb, Ni, Pd, Pt, Re, Ru, Si, Sn, Ta, V, W, Y, Zr

### $\beta$ Ti-X

- SQS @ 25, 50, 75 at.% X (20 \* 3 = 60 structures)
- Supercell @ 25, 50 at.% X (20 \* 2 = 40 structures)

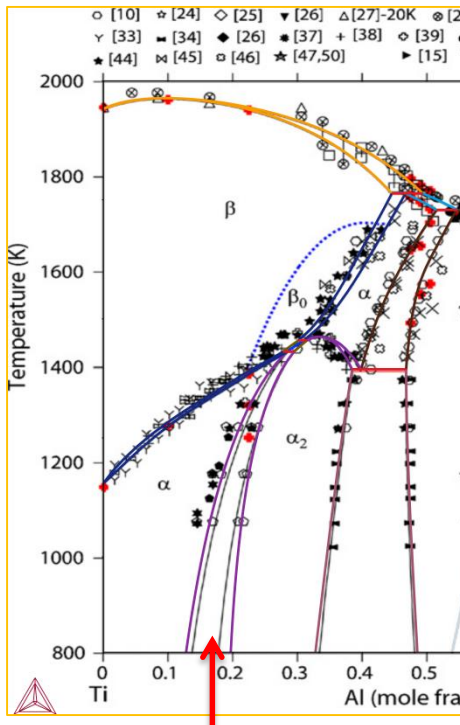
### $\alpha$ Ti-X

- SQS @ 25 at.% X (19 structures)
- Supercell @ 25, 50, 75 at.% X (57 structures)

### $\omega$ Ti-X (next step)

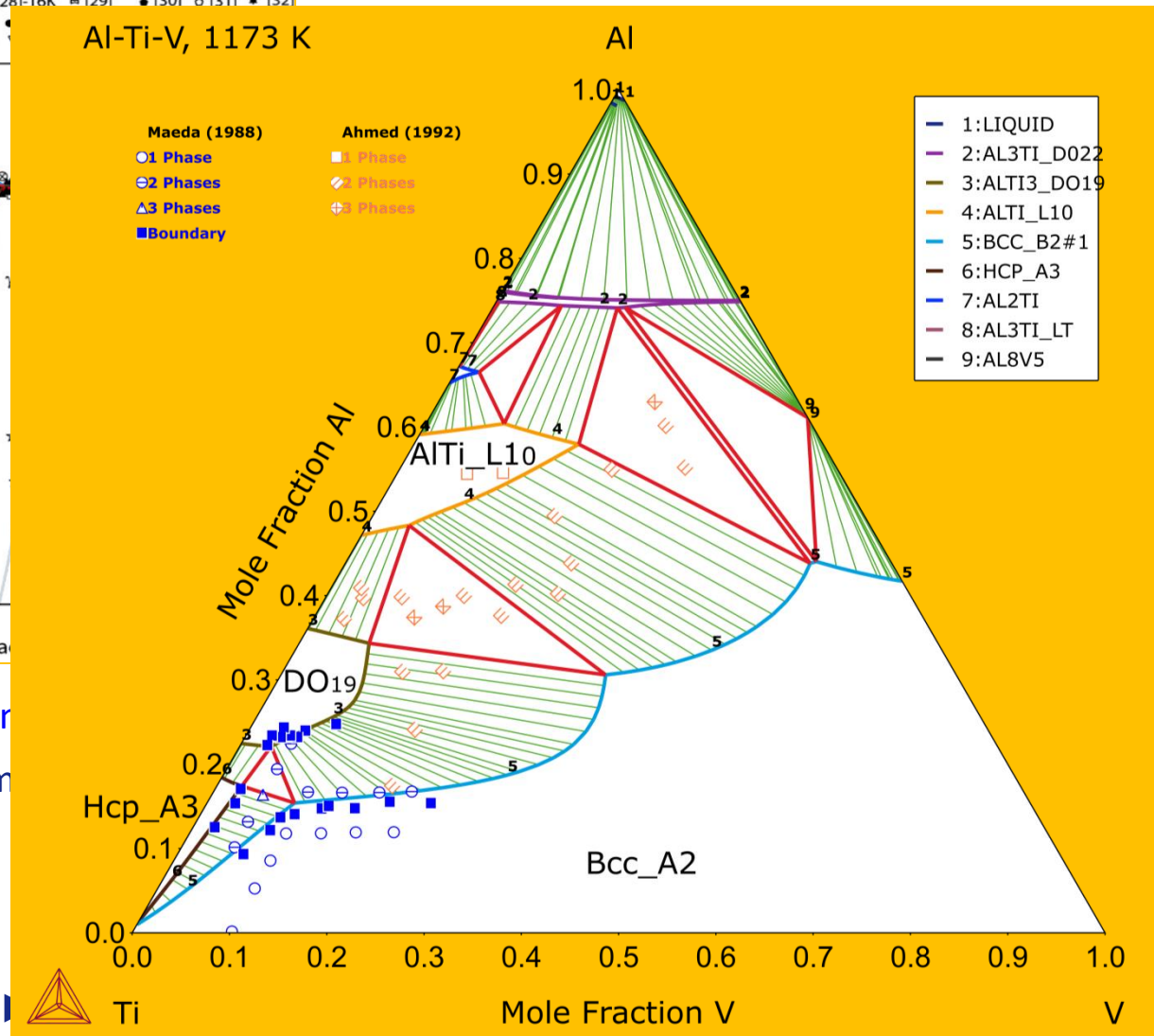
- SQS Ti<sub>36</sub>X<sub>12</sub> (21 structures)
- Supercell Ti<sub>96</sub>X<sub>12</sub>, Ti<sub>81</sub>X<sub>27</sub>, Ti<sub>54</sub>X<sub>54</sub> for X  $\neq$  Z
- Supercell Ti<sub>96</sub>X<sub>12</sub>, Ti<sub>81</sub>X<sub>27</sub>, for Z=Cr, Mo, Nb, Ta, V, W

## Example: Al-Ti-V



Al-Ti with modification on  
Witusiewicz, J Alloys Com

Al-Ti -V, this work



# Thank You!

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